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more stable by participating in a transfer of electrons or by sharing electrons with other atoms. In nuclear reactions, it is the nucleus of the atom that gains stability by undergoing a change of some kind.

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Chapter 10-5 10.15 If an artifact has 1/8 of the amount of C-14 compared to living organisms, it has decayed by three half-lives ($\frac{1}{2} \times \frac{1}{2} \times \frac{1}{2}$). 1 half-life 5,730 years 3 half-lives $\times = 17,200$ years 10.16 Use the amount of radioactivity (mCi/mL) as a conversion factor to convert the dose of radioactivity from millicuries to a volume in milliliters.

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Chapter 23 Nuclear Chemistry Notes 1 CHAPTER 23 NUCLEAR CHEMISTRY 23.1 THE NATURE OF NUCLEAR REACTIONS radioactivity - the spontaneous decay of an unstable nucleus with accompanying emission of radiation. nuclide - atom with a specific number of protons and neutrons in its nucleus. \Rightarrow There are 271 stable nuclides in nature, others are radioactive

CHAPTER 23 NUCLEAR CHEMISTRY

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