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kathrynphillips4. Chemistry Chapter 3: Scientific Measurement. accuracy, precision, error, percent error, the closeness of a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be.... a measurement to the true value of what is be..... a measurement to the true value of what is be.... a measurement to the t

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#### **Scientific Measurement**

Chapter Test A - Campbell County Schools. Chapter 3 Scientific Measurement 73 37. Calculate the density of a liquid that has a mass of 14.0 g and a volume of 18.0 cm3. ... Chapter 3 Scientific Measurement 57 Section Review ... Make the following conversions using Tables 3.1 and 3.2. Write your answers in scientific notation. a. 125 g to ...

# Core Teaching Resources Chemistry Answers Chapter 3 ...

SCIENTIFIC MEASUREMENT Class Column B density precision conversion factor Chapter Test B A. Matching Match each term in Column A write the letter of the correct term on the line. 2. 3. 8. 10. Column A how close a single measurement comes to the actual value of whatever is being measured

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# Test Review key - Chapter 3 Test Review Homework Name Hour ...

\_\_C\_\_ Measure of how close a series of measurements are to one another a. accuracy 2. \_\_A\_\_ Measure of how close a measurement that are known plus one that is estimated c. precision 4. \_\_E\_\_ A value determined in the laboratory d. scientific notation 5

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1. Put in correct scientific notation. 2473000 0.006520 2. The true value for a measurement is 24.36 m. The experimental value is 20.50 m.

# Review Questions for Test - Chapter 3 (answers are below)

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3.1 Using and Expressing Measurements > ... 1

# 3.1 Using and Expressing Measurements >

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